# Imagine the Future Power Plastic

Nano-morphology and charge photo-generation in oBHJs based on fullerene and bridged dithiophenes with Carbon and Silicon bridging atoms

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## **Conversion of Sunlight into Electricity**

**Three step process** 

- 1. Absorption of photons in a material
- 2. Photo-induced charge separation
- 3. Collection of charges at electrodes

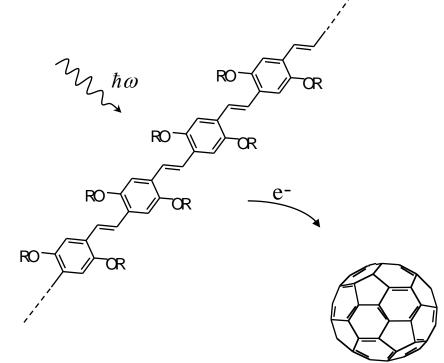


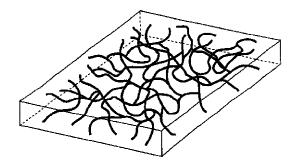
*"inks" ---- with electronic functionality!* 

The Problem: Semiconducting polymers cast from solution have low mobility with recombination lengths in the 10-20 nm range



### The "Idea" ----



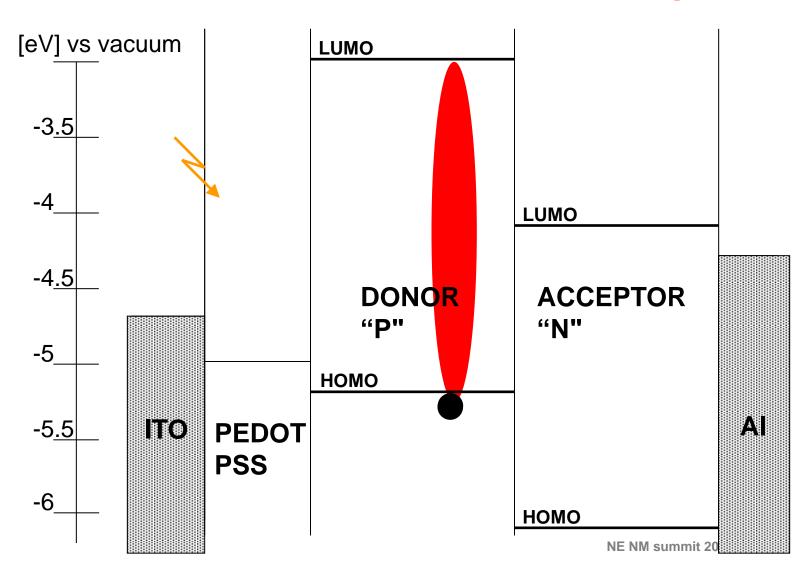


Self-Assembly of "Bulk Heterojunction" material by Spontaneous Phase Separation



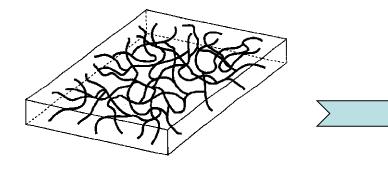
### Working principle of OPV

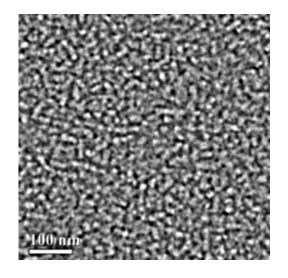
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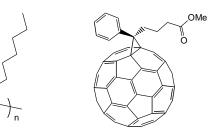




## Donor and Acceptor components form charge separating heterojunctions







**Bulk Heterojunction Material** 



PCBM

**Bicontinuous interpenetrating networks** 

## Self-assembled nanoscale material with charge-separating junctions everywhere!

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## Low Cost "Plastic" Solar Cells





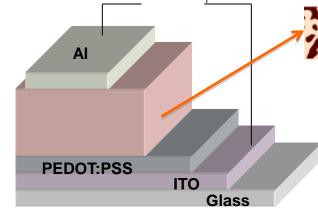
### Roll-to-roll manufacturing of plastic solar modules at Konarka facility in New Bedford, MA



# **Device architecture**

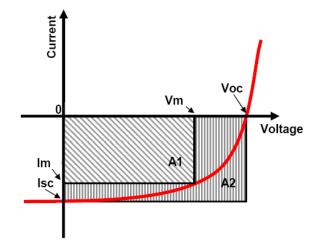
### Must break the symmetry ----

Use two different electrodes with different work functions. Electrons will automatically go toward lower work function contact and holes toward higher work function contact



Light

Thin Film of Phase Separated Bulk Heterojunction Material



### Need to improve the efficiency

PCE = Isc Voc FF / Power (input)



Voc =[Efull(LUMO) – Epol(HOMO)]

Synthesize polymers with deep HOMO New Acceptors ---- so that Voc  $\rightarrow$  Eg/e

Jsc

Smaller band gap Control nano-morphology

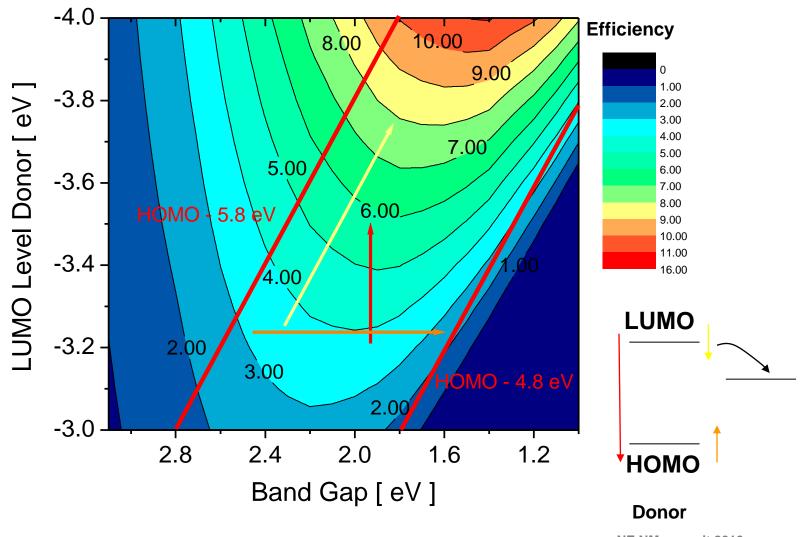
FF

Recombination Reduce density of interfacial traps and increase μτR



Working principle of OPV

**Efficiency Prediction OPV** 



Assuming fullerenes as acceptors

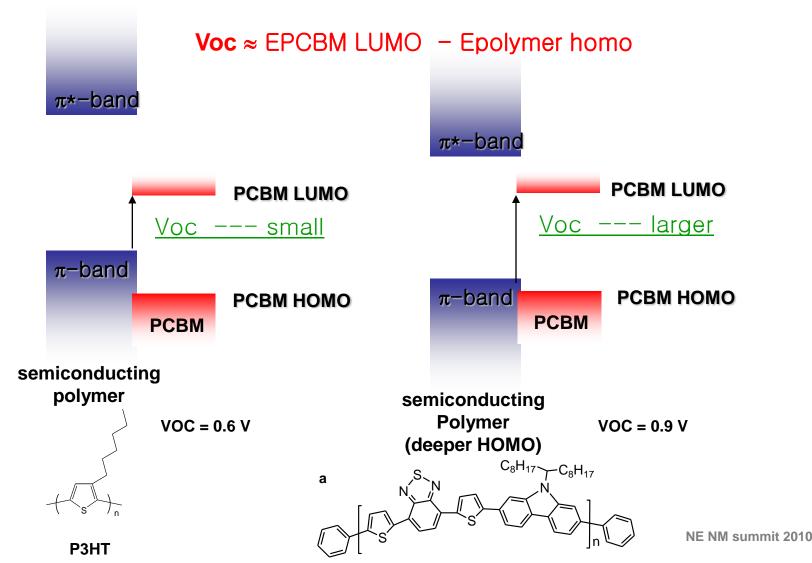
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## **Origin of Voc**

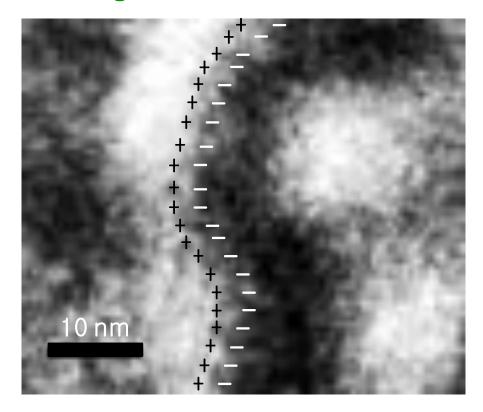
When irradiated with high light intensity,

Fermi levels must be equal (holes in the  $\pi$ -band and electrons in PCBM LUMO)





### Light on --- at V =Voc

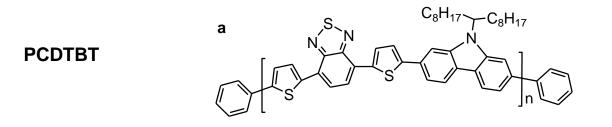


VOC  $\approx$  Ea(LUMO) – Ed (HOMO)

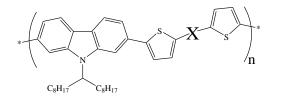
 $\Delta$ V"capacitor" = Voc

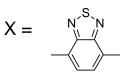


poly[N-9"-hepta-decanyl-2,7-carbazole-alt-5,5-(4',7'-di-2-thienyl-2',1',3'-benzothiadiazole)



#### **Flexible Chemistry**



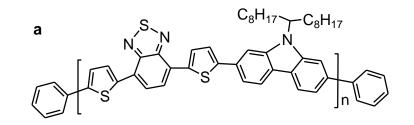


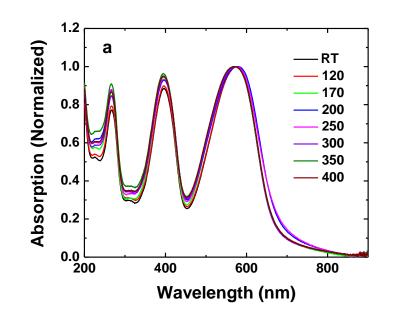
Mario Leclerc and colleagues Université Laval Quebec City

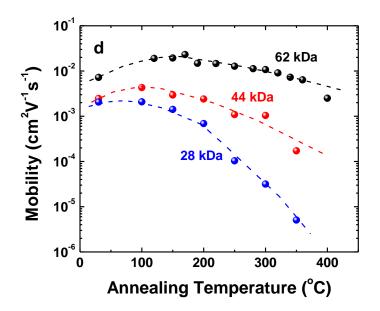
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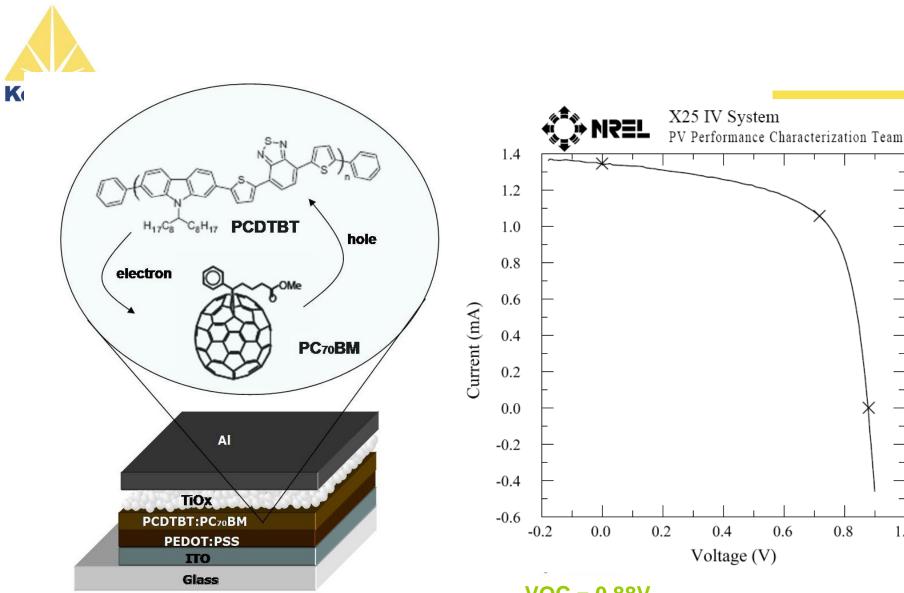


### **PCDTBT:** <u>Stable</u> Semiconducting Polymer









VOC = 0.88V FF = 64% Power Conversion Efficiency = 6 %

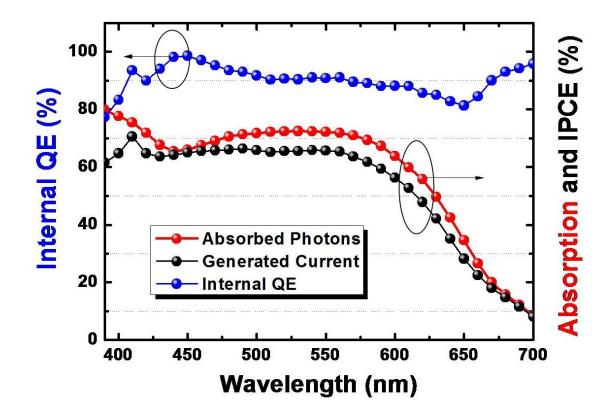
Sung Heum Park

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1.0



## **Potential for high efficiency**



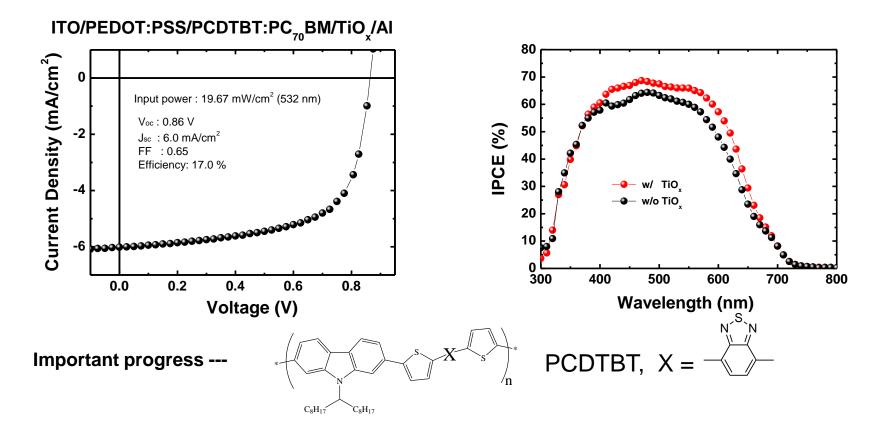
### Internal Quantum efficiency approaches 100%.

Nearly every photon yields one "e-h" polaron pair and all photo-generated carriers are collected at the electrodes.



## **17% Power Conversion Efficiency**

### for wavelengths within the absorption band

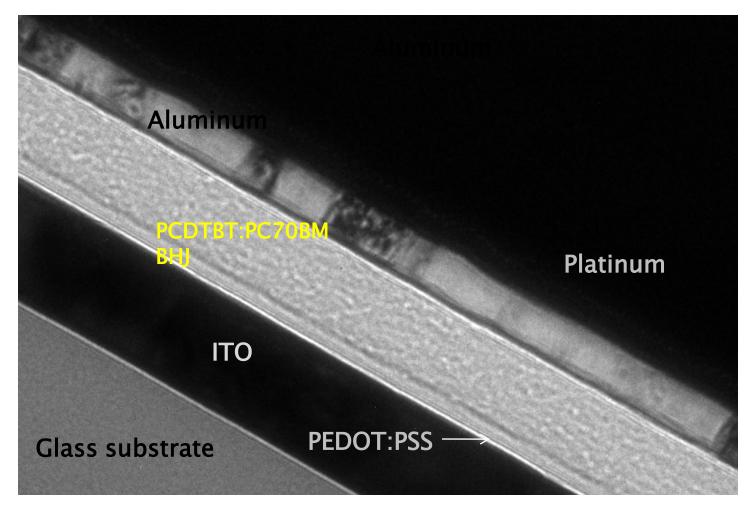


### Demonstrates the potential of BHJ technology for high PCE

- Reduce the band gap
- Increase the FF ( $\mu\tau$ -product).



### Cross-sectional TEM Image of the PCDTBT:PC70BM BHJ



PCDTBT:PC70BM=1:4 Image was obtained under defocus

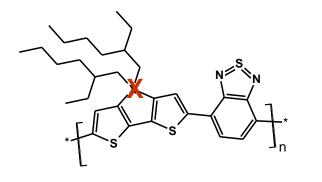


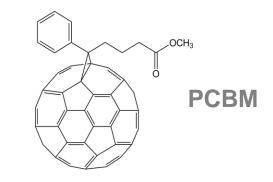
# It's all about Morphology



## **Control of morphology a case study**

Alternating D/A low-bandgap copolymers: dithiophenes-benzothiadiazole





X=Si X=C (PCPDTBT) <u>"Si-bridged (Si-b)"</u> VS. "C-bridged (C-b)"

A case of study because:

- Minimal structural changes
- Maximum impact on oBHJ nanomorphology

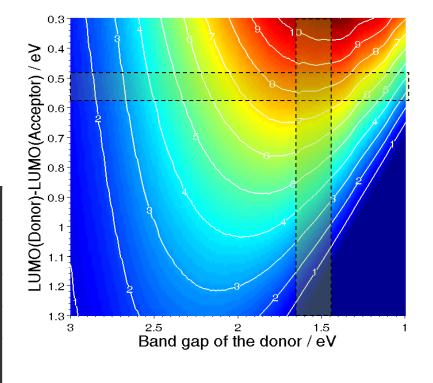
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## **Power conversion: potential**

	НОМО	LUMO	Vbi (HOMO <sub>poly</sub> -LUMO <sub>РСВМ</sub> )
C-b	-5.3 eV	-3.55 eV	1.3V
Si-b	-5.3 eV	-3.6 eV	1.3V

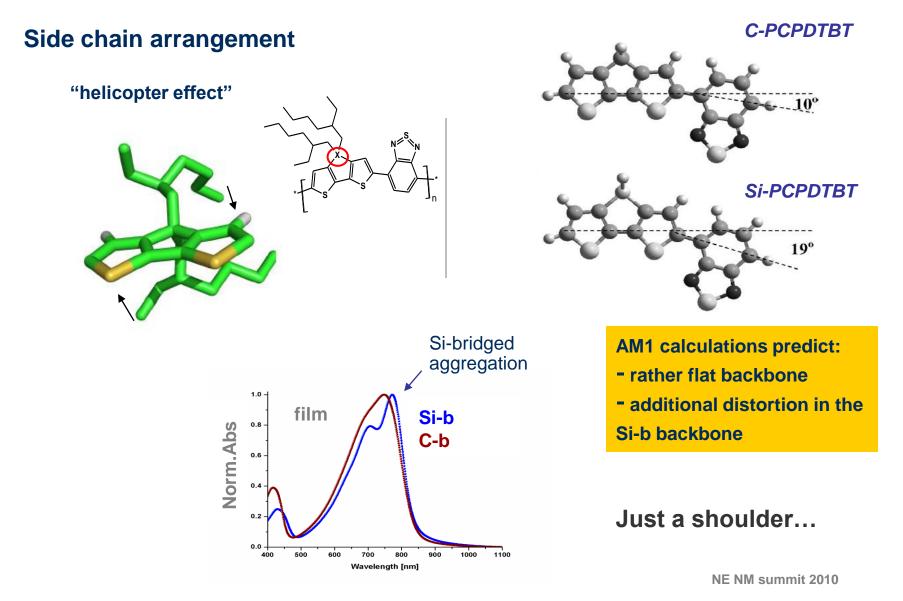
OFETs	Polymer	Blend (1:2)	
	μ <sub>h</sub> [cm²/Vs]	μ <sub>h</sub> [cm²/Vs]	μ <sub>e</sub> [cm²/Vs]
C-b	5x10 <sup>-3</sup>	3x10 <sup>-4</sup>	4x10 <sup>-4</sup>
Si-b	10 <sup>-2</sup>	<b>10</b> <sup>-3</sup>	<b>10</b> <sup>-3</sup>
rr-P3HT	10 <sup>-3</sup>	5x10 <sup>-4</sup> (1:1)	4x10 <sup>-4</sup> (1:1)



Assuming "full" Voc, FF=0.65, max EQE=65%

Max estimated PCE in the range 7-8%



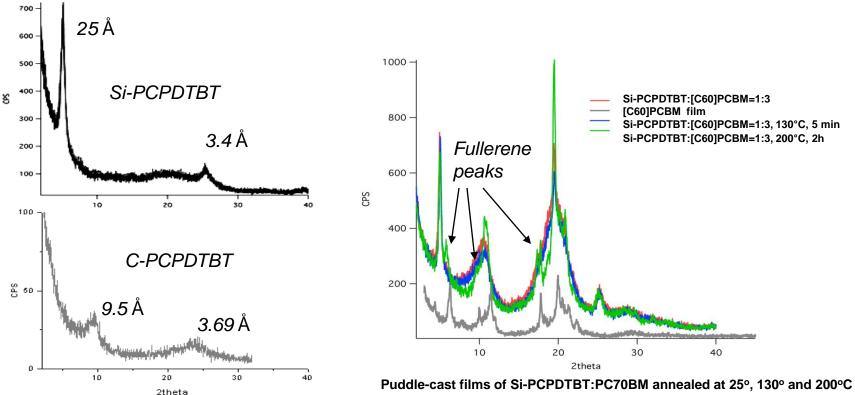




### X-Ray diffraction analysis

• Enhanced pi-pi stacking, in Si-PCPDTBT, the stacking distance (3.4 A) is <u>shorter</u> than what found in C-PCPDTBT and P3HT.

• Neither the presence of PCBM nor annealing significantly affects the crystallinity of the polymer.

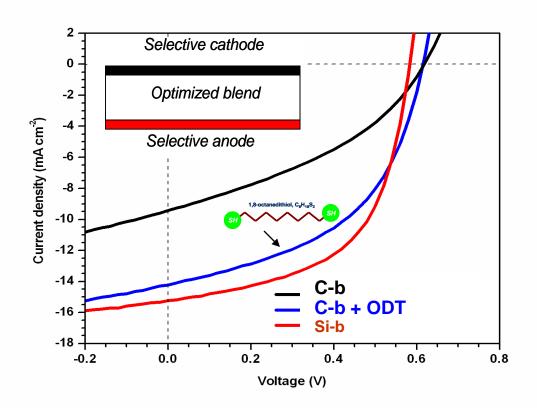


Films of pristine Si-PCPDTBT as cast and annealed (red line).

ok, some crystallinity.. NE NM summit 2010



## **Photovoltaic devices**



 C-b: solar cells dominated by losses
 ODT: reduces losses (& bad flavour..) <u>The effect seems to be partial!</u>
 Si-b: ΔJsc ~ +6mA/cm<sup>2</sup>

**∆FF ~ +50%** 

### we LOOK INTO...

	Jsc (mA/cm^2)	FF	Voc (V)	Eff(%) corrected
C-b	9.4	0.37	0.62	2.2 (up to 3.5%)
C-b + ODT	14.2	0.49	0.62	4.3 (up to 5%+)
Si-b	15.2	0.57	0.58	5 (up to 6%+)

**2D TEM: overlay** 

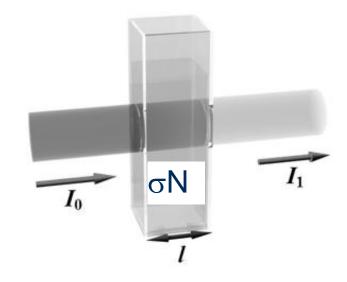
**Ideal TEM Bright-Field mode** 

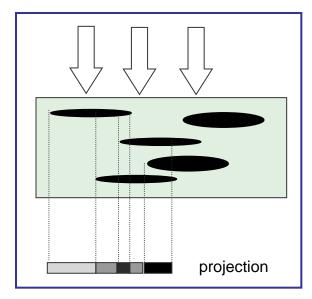
Beer-Lambert law

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 $T = I/I_0 = 10^{-\sigma NI}$ 

σ=absorption cross sectionN=density of absorbersI=film thickness





Generally: density<sub>polymer</sub> <density<sub>fullerene</sub>

 Ideally BF TEM should produce a grey scale image with local intensity reduced by the number of fullerene molecules along the beam path

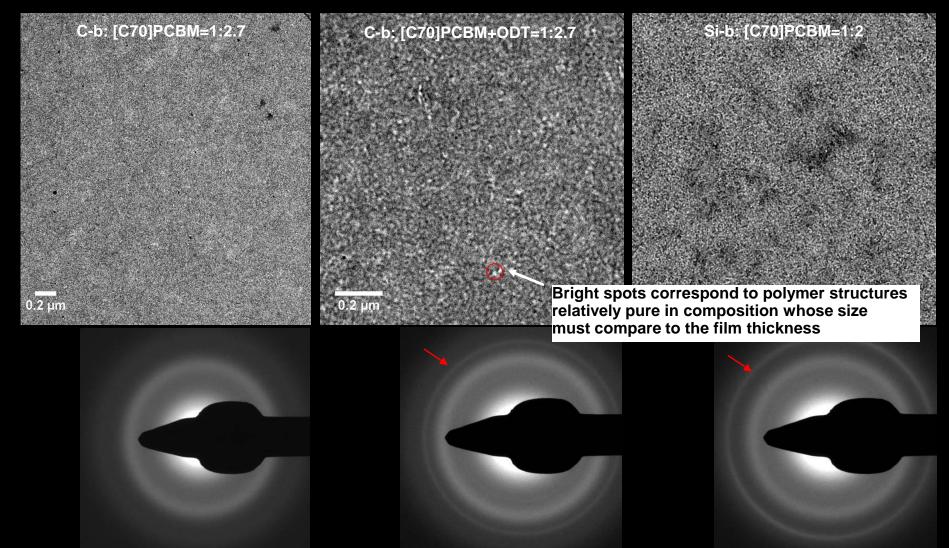
 ratio between the fullerene size/ film thickness determines the relatively low contrast in highly intermixed systems

Diffraction contrast due to crystallinity may add



### **Bright Field TEM**

Films prepared as for solar cells on PEDOT:PSS, thickness 90-120nm



Rms roughness of all films 1<r<5nm

TEM microscope resolution < 1nm

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### **BF-TEM - DFT analysis**

### Typical image size: 1024x1024 pixels

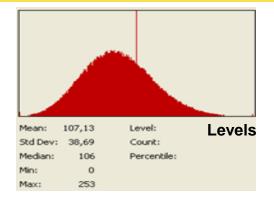
### 256 grey levels (8 bits)

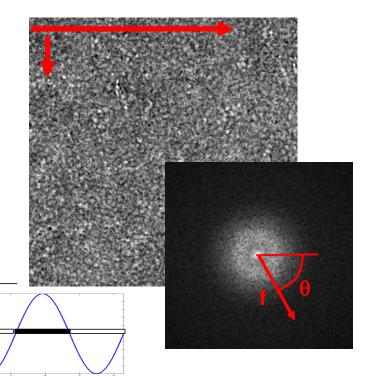
Human eye can distinguish <100 grey levels (maybe ties on white shirt under ideal illumination..)

period

*l*(*x*,*y*)=amplitude contrast intensity

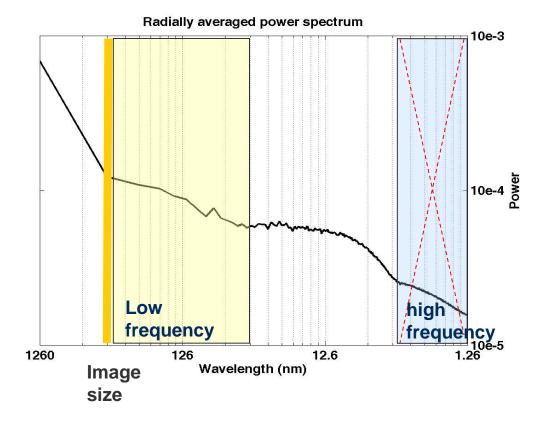
$$PSD(f_x, f_y) = \frac{1}{L^2} \left| \int_0^L \int_0^L e^{-i2\pi x f_x} e^{-i2\pi y f_y} I(x, y) dx dy \right|^2$$
$$PSD(f) = \frac{1}{2\pi} \int_0^{2\pi} PSD(f, \theta) d\theta$$







### Si-b:PCBM



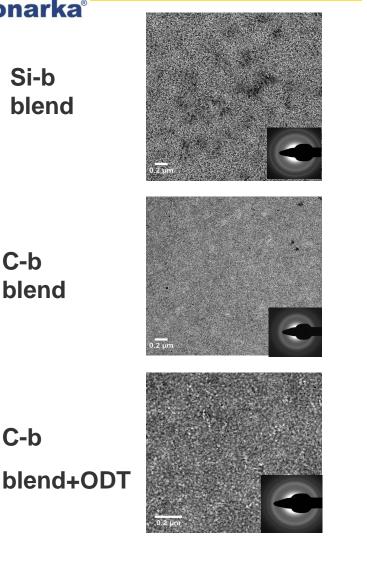
### **PSD** analysis **SFS = Smallest Feature Size**

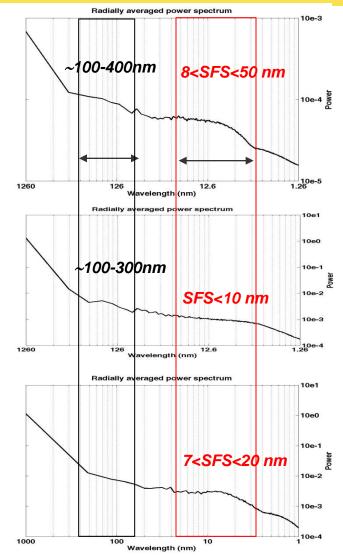
Si-b blend

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C-b blend

C-b





Meso and nanoscale phase separation

Molecularly dispersed

Regular features, no mesoscale

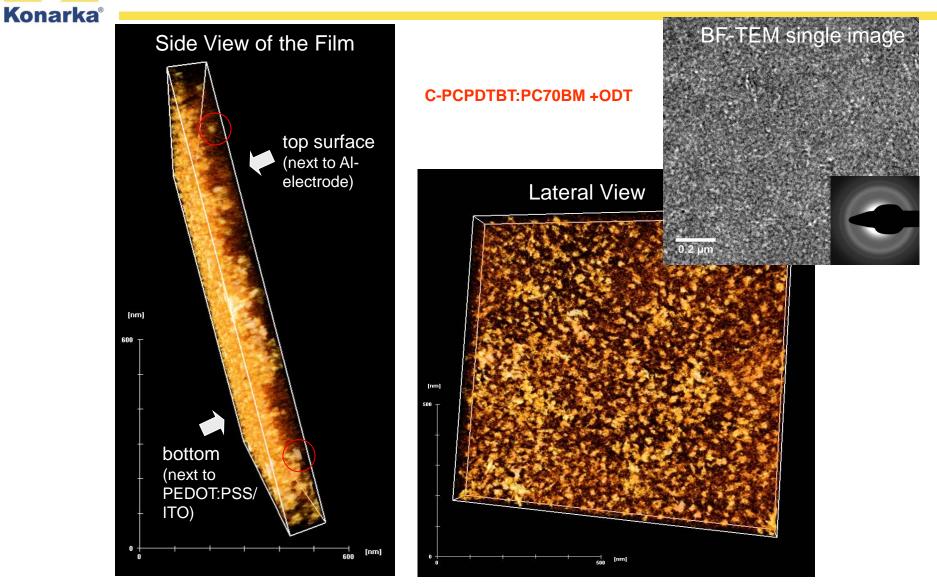
But low resolution images do not help...



### What can we deduce from TEM/PSD analysis?

- Iarge features in TEM have generally depth comparable to the lateral size
- C-b blends tend to be amorphous and molecularly dispersed even at high fullerene loads, FS<10nm
- $\cdot$  ODT induces phase separation and produces regularly distributed structures with charactersitic size of FS  $\sim$  10-20nm
- Si-b blends phase separate at the meso and nanoscale, with averaage FS >>10nm

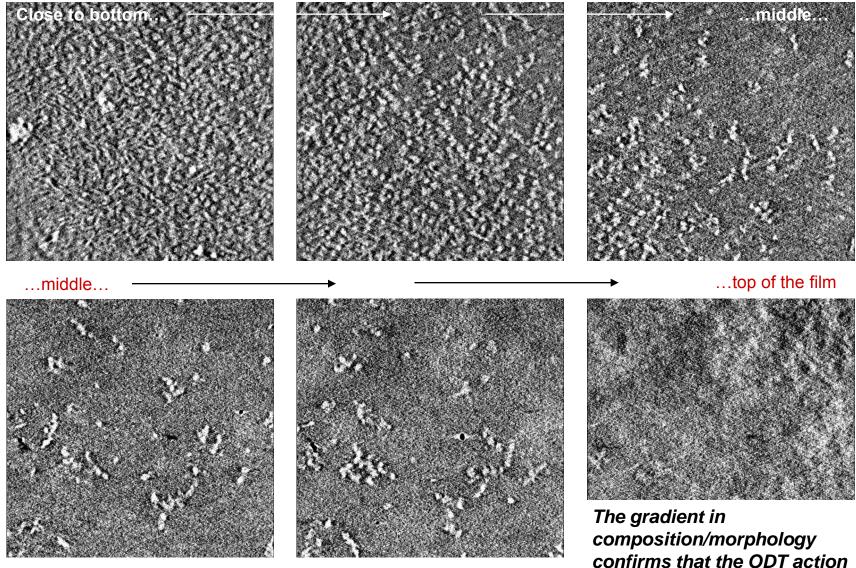
Feature size: C-b < C-b+ODT< Si-b TEM Tomography



Polymer-rich areas are visualized as bright orange spots



### Outcome of electron tomography: slices out of the 3D data stack

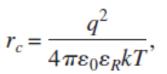


is incomplete across the film

Polymer-rich areas look brighter. Slice dimensions are 1082 nm x 1082 nm.



For  $\epsilon$ =3-4, the capture radius ~ 16nm



SFS = Smallest Feature Size	Max EQE	
SFS C-b << r <sub>c</sub>	35%	
SFS C-b +ODT ~ $r_c$	55%	
SFS in Si-b is distributed around $\rm r_{c}$	63%	

- Losses in C-b blends appear related to the isolated fullerene molecules/nanoclusters

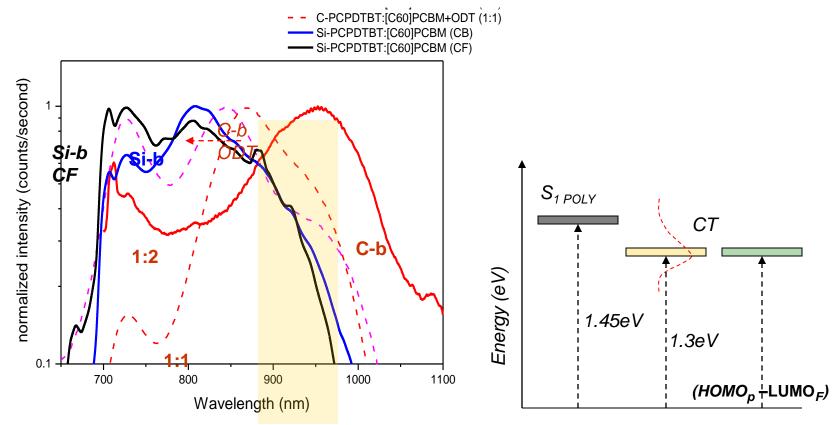
- in C-b blends, ODT helps the fullerene to demix leading to the formation of small polymer-rich clusters

- Si-b forms an optimal phase separation due to a strong tendency to aggregate/crystallize and to a favourable hierarchy of domain size

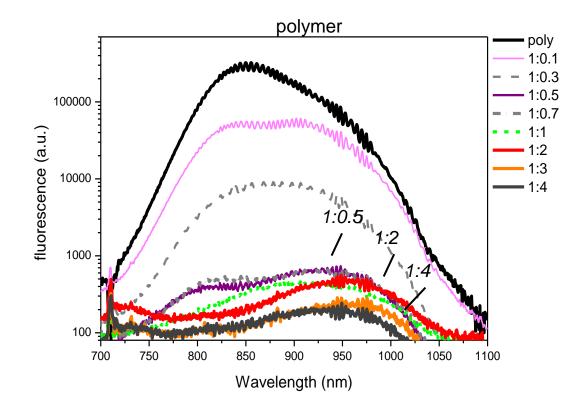
# **CT** emission

- An emissive CT (1.3eV) is present as decay path in C-b. The width of the emission correlates to a distribution of CT energies related to D/A distances
- films prepared with ODT show a quenched CT emission, quenching increases with the fullerene content
- The main emission peak of CTC is increasingly blue-shifted by adding fullerene

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C-b: compositional dependence of CT emission



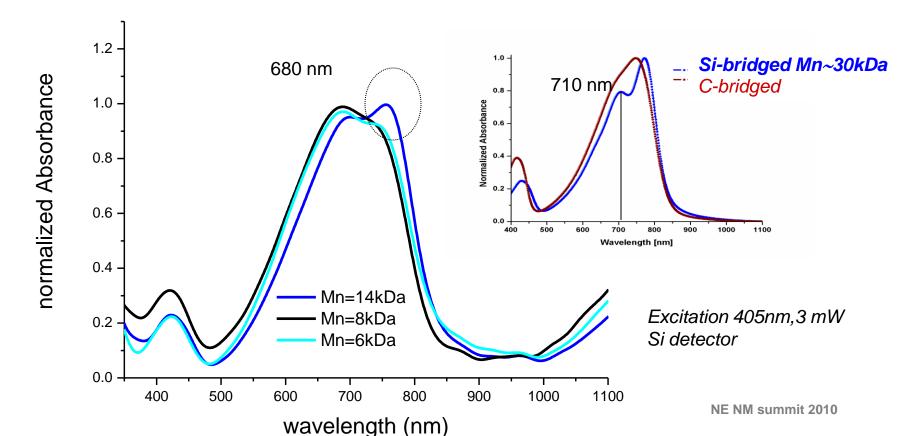
in C-b high quenching
of the main polymer
peak is required to
observe CT

• at high fullerene content the quenching stays complete



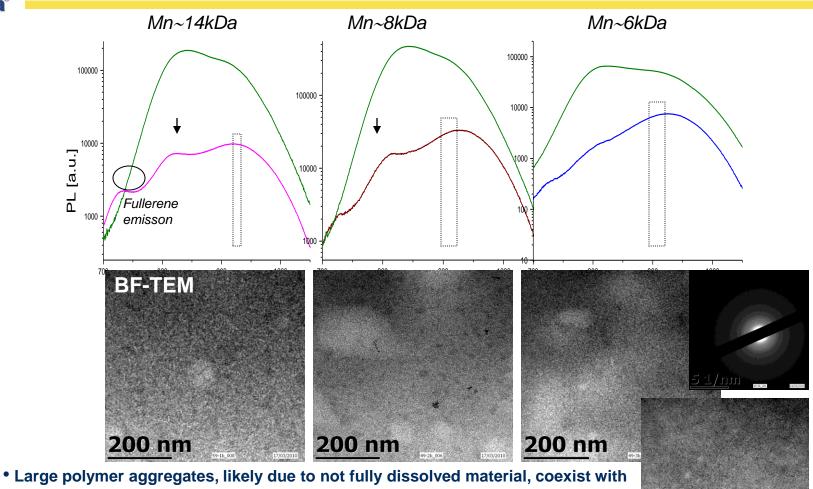
### Nano-morphology vs. moelcular weight in Si-bridged blends

Si-bridged	Mn (Da)	Mw (Da)	PDI
low-1	14324	20357	1.42
low-2	8047	12931	1.61
low-3	6061	9598	1.58





### Low molecular Si-b blends (1:2)



- a highly intermixed morphology
- The quenching of the main polymer emission peak is between 10 and 100 times.
- PCBM emission disappears by reducing Mn. All blends appear amorphous by Electron Diffraction.
- The polymer emission peak at 830 decreases by reducing Mn, revealing a well defined peak at 930-950nm

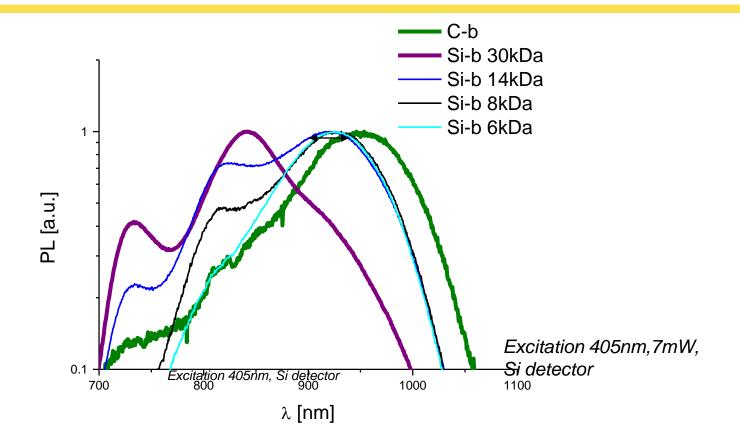
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200 nm

**Pristine polymer** 

### Nano-morphology vs. moelcular weight in Si-bridged blends

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• by reducing Mn the main polymer emission is progressively quenched, together with the fullerene emission at 710nm



## Conclusions

• The morphological analysis of Si-bridged:[C70]PCBM in comparison to its carbon-bridged analogous leads to correlate the improved photogeneration to slightly larger phase separation.

• Films processed under optimized conditions show in TEM a feature size in the length scale in the range of the exciton diffusion length.

• The higher chain packing and crystallinity of the Si-bridged copolymer is thought to be the driving force for the fullerene demixing

• CT emission in C-PCPDTBT blends shows relation to the losses affecting the system, and may be an indicator of the degree of intermixing in the blend.

• An NIR emission peak was found in well intermixed Si-bridged blends based on very low molecular weight material, indicating a close relation between intermixing and infrared emission



